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like in hydrogen, that has only one electron in the orbital. a full octet wouldn't affect the activity of these metals. (if anyone has a different answer please send it to me) why does hydrogen act like a group 17 element when forming covalent bonds. its not stable by itself so it bonds quickly.

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concepts. Terms in this set (68) valence  
electron. the electrons on an atom that  
can be gained or lost in a chemical  
reaction. covalent bond. neutral atoms  
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Edit. Edit. ... answer choices . one atom of chlorine. two atoms of chlorine. one atom of oxygen.

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1. Use the full name of the cation (first ion) 2. Add the suffix -ide to the name of the anion (second ion) if it is monatomic (Use the full name of the anion if it is polyatomic) \*Include the roman numeral if there is a transition metal. Old System. -ous for lower charged ion, -ic for higher

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charged ion.

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PLAY. Match. Gravity. Created by.  
emma\_pear. Terms in this set (32)  
Identify and define the three major types  
of chemical bonding. In ionic bonding,  
large numbers of oppositely charged

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ions join because of mutual electrical

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\_\_\_\_ 6. A repetition of properties occurs  
when elements are arranged in order of  
increasing atomic number. \_\_\_\_ 7.  
vertical column in the periodic table

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Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 8. Which of the following elements is in the same period as phosphorus? a. magnesium c. nitrogen

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halogen atom bonded to an  $sp^3$   
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