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Objectives 1. Define pure substances and mixtures. Pure substances - type of matter with a fixed composition, element or compound (composition is definite) Mixtures - 2 or more substances that can be separated by physical means (Composition is variable) 2.

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**Chapter 18 Acids
And Bases Study
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Ideas Answers 14.

Because protons and
neutrons reside in the
nucleus of atoms, they
are called nucleons.

16. There are two
forces among the
particles within the
nucleus. The first,
called the electrostatic
force, is the force

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Chapter 18

between electrically charged particles. The second force,

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OpenStax 1. The alkali metals all have a single s electron in their outermost shell. In contrast, the alkaline

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earth metals have a completed s subshell in their outermost shell.

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18.1 Introduction to

Acids and Bases pages

634-643 Practice

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Problems pages
635-640 1. Write
balanced equations for
reactions between the
following. a. aluminum
and sulfuric acid $2\text{Al}(s)$
 $+ 3\text{H}_2\text{SO}_4(aq) \rightarrow 0 \text{ Al}$
 $2(\text{SO}_4)$

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This Study Guide for Content Mastery for Chemistry: Matter and Change will help you learn more easily from your textbook. Each textbook chapter has six study guide pages of questions and exercises for you to complete as you read the text. The study guide pages are divided into sections that match those in your text.

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