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4 Cl(s). How many grams of NH 3 (g) reacted? Is the law of conservation of mass observed in the above reaction? Justify your answer. 157.5 g 106.5 g 51 g Yes. Mass of reactants equals mass of products. Section 3.2 Assessment page 79 10. Classify each example as a physical change or a chemical change. a. crushing an aluminum can physical

**Matter—Properties and ChangesMatter—Properties and Changes**  
Chemistry - Chapter 9 Assessment Answers 9.1 Assessment 3. The positive charge of a group A metal is equal to its group number. The charge of a group A nonmetal is the group number minus 8. 4. from the number of electrons lost 5. -ite or -ate 6. Group 1A metals, 1+; Group 3A (aluminum), 3+ Group 5A nonmetals, 37.

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